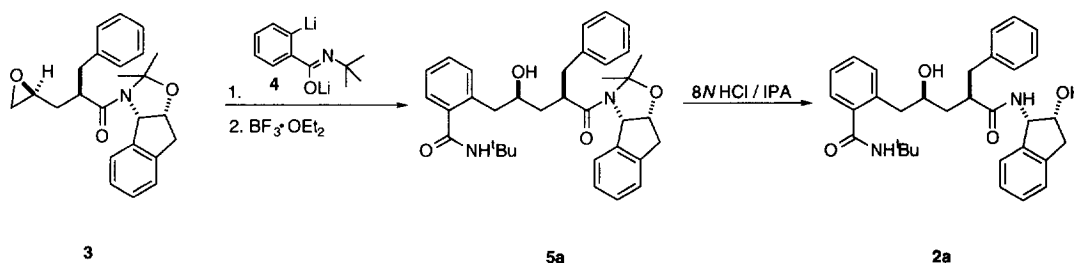


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2-arylethanols in which the *S*-configuration of the hydroxyl group would be derived from the epoxide prior to the introduction of the aromatic P<sub>1</sub> group. The success of the reaction scheme would then be based on a Lewis acid mediated opening of epoxide **3** by an appropriately substituted aromatic nucleophile.<sup>7</sup>

Scheme 1



Directed metallation of *N*-*tert*-butylbenzamide (2 equivalents *n*-BuLi/0 °C/2 h) afforded dilithio anion **4**, which was treated with a solution of epoxide **3** at -70 °C followed *immediately* by the addition of 1 equivalent of  $\text{BF}_3 \cdot \text{OEt}_2$ . After work up the phenethyl alcohol adduct **5a** was isolated in good yield without any detectable racemization. Hydrolysis of the ketal protecting group was accomplished by the action of 8*N* HCl in isopropanol to afford title compound **2a** as a crystalline solid.<sup>8</sup>

Table 1: Effect of P<sub>1</sub> Replacement

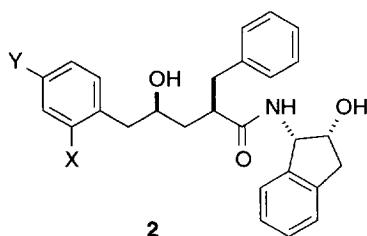
Compound	R	IC <sub>50</sub> (nM)	CIC <sub>95</sub> (nM)
<b>6</b>		80	nd <sup>1</sup>
<b>7</b>		38	3,000
<b>2a</b>		4.7	800

<sup>1</sup>not determined

Direct comparison of the aromatic P<sub>1</sub> inhibitor **2a** to several saturated cyclic isosteres revealed that this modification was favorable in both the *in vitro* assay and the whole cell assay (Table 1).<sup>9</sup> Encouraged by these results, we examined a number of substituted  $\beta$ -arylethanol inhibitors.

The short and efficient reaction scheme described above served well for the synthesis of a wide variety of arylethanol inhibitors that contained the benzyl indanamide subunit (Table 2). Furthermore, we were able to convert some of the final compounds into a variety of modified P<sub>2</sub> derivatives. For example, anisole **2d** could be cleanly demethylated (BBr<sub>3</sub>/DCM/rt) to afford phenol **2c** in 56% yield. Oxidation of thioanisole **2i** with 1.5 equivalents of Oxone<sup>®</sup> provided sulfone **2j**. Finally, dealkylation of *tert*-butylsulfonamide **2l** with 8N HCl in isopropanol provided access to primary sulfonamide **2k**.

**Table 2: Aromatic P<sub>1</sub> Analogs**



Compound	X	Y	IC <sub>50</sub> (nM)
<b>2b</b>	H	H	1,200
<b>2c</b>	OH	H	1,600
<b>2d</b>	OCH <sub>3</sub>	H	80
<b>2e</b>	OCH <sub>3</sub>	OCH <sub>3</sub>	105
<b>2f</b>	OCH <sub>3</sub>	OPh	45
<b>2g</b>	O <sup>t</sup> Bu	H	97
<b>2h</b>	OCH <sub>2</sub> OCH <sub>3</sub>	H	276
<b>2i</b>	SCH <sub>3</sub>	H	95
<b>2j</b>	SO <sub>2</sub> CH <sub>3</sub>	H	314
<b>2k</b>	SO <sub>2</sub> NH <sub>2</sub>	H	290
<b>2l</b>	SO <sub>2</sub> NH <sup>t</sup> Bu	H	6
<b>2m</b>	CONH <sup>t</sup> Bu	CH <sub>3</sub>	1.7

The above compounds were evaluated for their ability to inhibit the HIV PR enzyme and the results are shown in table 2. Interestingly, the aryl ether and thioether compounds (**2d-2i**) were found to be fairly potent inhibitors *in vitro* ( $IC_{50} < 100$  nM). However, phenol **2c** was 20-fold less active than its corresponding anisole analog. Sulfonamide **2l** and carboxamide **2a** were equipotent but removal of the lipophilic *t*-butyl group resulted in a significant drop in potency as evidenced in compound **2k**. Substitution of a methyl group *meta* to the carboxamide moiety gave the most potent compound thus far (**2m**;  $K_i = 1.7$  nM), which was also reflected in whole cell assay ( $CIC_{95} = 400$  nM). In conclusion, we have demonstrated that substitution of an aromatic group in  $P_1/P_2$  of CRXIVAN® can lead to highly active HIV PR inhibitors.

**Acknowledgements:** We thank Dr. Paul Darke and Joan Zugay for assay work. We are grateful to Dr. H.G. Ramjit, Mr. A. B. Coddington, Mr. Matthew Zrada, Mr. Ken Anderson and Ms. Patrice Ciecko for analytical support. Ms. J. F. Kaysen is gratefully acknowledged for the manuscript preparation.

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(Received in USA 24 June 1996; accepted 16 July 1996)